

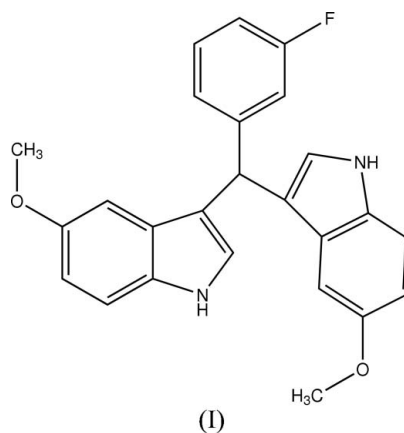
5,5'-Dimethoxy-3,3'-(3-fluorophenyl-
methanediyl)bis(1*H*-indole)Shi-Gui Tang,^a Dong-Mei
Zhang,^b Wen-Yuan Wu,^b Liu
Shan^b and Cheng Guo^{b*}^aCollege of Life Sciences and Pharmaceuticals,
Nanjing University of Technology, Xinmofan
Road No. 5 Nanjing, Nanjing 210009, People's
Republic of China, and ^bDepartment of Applied
Chemistry, College of Science, Nanjing
University of Technology, Xinmofan Road No. 5
Nanjing, Nanjing 210009, People's Republic of
China

Correspondence e-mail: guocheng@njut.edu.cn

Key indicators

Single-crystal X-ray study
 $T = 293$ K
Mean $\sigma(\text{C}-\text{C}) = 0.004$ Å
 R factor = 0.061
 wR factor = 0.179
Data-to-parameter ratio = 14.9For details of how these key indicators were
automatically derived from the article, see
<http://journals.iucr.org/e>.The title compound, $\text{C}_{25}\text{H}_{21}\text{FN}_2\text{O}_2$, was synthesized by the
reaction of 5-methoxy-1*H*-indole and 3-fluorobenzaldehyde in
ethanol, using CuBr_2 as a catalyst under microwave irradiation.
In the crystal structure, there are two intermolecular
hydrogen bonds, one $\text{N}-\text{H}\cdots\text{O}$ and one $\text{C}-\text{H}\cdots\text{F}$, also one
intermolecular $\text{N}-\text{H}\cdots\pi(\text{arene})$ contact.

Comment

Development of bis(indolyl)alkane synthesis has been of
considerable interest because of the wide occurrence of bis-
(indolyl)alkanes in various natural products possessing
biological activity (Bell *et al.*, 1994) and their usefulness for
drug design. We report here the crystal structure of the title
compound, (I).The molecular structure of compound (I) is shown in Fig. 1.
Bond lengths and angles are within normal ranges (Allen *et al.*,
1987). Ring *A* comprises atoms C20–C25, ring *B* atoms C5/N1/
C6/C7/C8 and ring *C* atoms C14/N2/C15/C16/C17. The various
dihedral angles are $A/C = 109.4$ (2)°, $B/C = 94.4$ (1)° and $A/B =$
 80.0 (2)°.The crystal structure of (I) is stabilized by two inter-
molecular $\text{N}-\text{H}\cdots\text{O}$ and $\text{C}-\text{H}\cdots\text{F}$ hydrogen bonds and one
intermolecular $\text{N}-\text{H}\cdots\pi(\text{arene})$ contact (Fig. 2 and Table 1;
 Cg1 is the centroid of atoms C11–C14/C17/C18). The $\text{N}-$
 $\text{H}\cdots\text{O}$ hydrogen bond and $\text{N}-\text{H}\cdots\pi(\text{arene})$ contact are also
present in a very similar compound, in which the fluoro group
is replaced by a nitro group (Guo *et al.*, 2006).

Experimental

Compound (I) was prepared by the reaction of 5-methoxy-1*H*-indole
(20 mmol) with 3-fluorobenzaldehyde (10 mmol) in ethanol (5 ml),
using CuBr_2 (0.446 g) as catalyst under microwave irradiationReceived 11 September 2006
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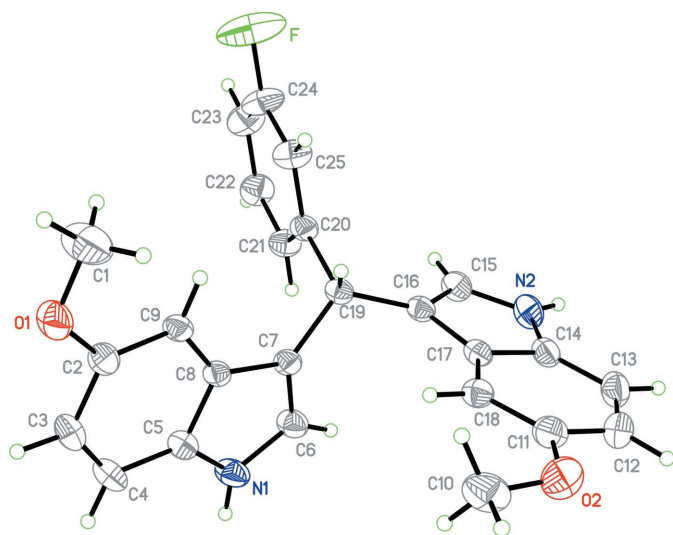


Figure 1
The molecular structure of (I), showing the atom-labelling scheme. Displacement ellipsoids are drawn at the 30% probability level.

(280 W) for 10 min. After completion, the reaction mixture was quenched with H₂O (10 ml) and extracted with EtOAc (3 × 10 ml). The combined organic layers were dried over Na₂SO₄, concentrated, and purified by column chromatography on SiO₂ (ethyl acetate–petroleum ether 1:3 v/v) to afford the pure product, (I) (m.p. 472–475 K). Crystals of (I) suitable for X-ray diffraction were obtained by slow evaporation of a methanol solution.

Crystal data

C₂₅H₂₁FN₂O₂ $Z = 4$
M_r = 400.44 $D_x = 1.284 \text{ Mg m}^{-3}$
 Monoclinic, *P*2₁/*n* Mo *K*α radiation
a = 9.1910 (18) Å $\mu = 0.09 \text{ mm}^{-1}$
b = 10.838 (2) Å $T = 293 \text{ (2) K}$
c = 20.879 (4) Å Block, colourless
 $\beta = 94.92 \text{ (3)}^\circ$ $0.40 \times 0.30 \times 0.30 \text{ mm}$
V = 2072.1 (7) Å³

Data collection

Enraf–Nonius CAD-4 4062 independent reflections
 diffractometer 2600 reflections with $I > 2\sigma(I)$
 $\omega/2\theta$ scans $R_{\text{int}} = 0.021$
 Absorption correction: ψ scan $\theta_{\text{max}} = 26.0^\circ$
 (North *et al.*, 1968) 3 standard reflections
 $T_{\text{min}} = 0.964$, $T_{\text{max}} = 0.976$ every 200 reflections
 4322 measured reflections intensity decay: none

Refinement

Refinement on F^2 $w = 1/[\sigma^2(F_o^2) + (0.0792P)^2 + 0.9173P]$
 $R[F^2 > 2\sigma(F^2)] = 0.061$ where $P = (F_o^2 + 2F_c^2)/3$
 $wR(F^2) = 0.179$ $(\Delta/\sigma)_{\text{max}} = 0.023$
 $S = 1.04$ $\Delta\rho_{\text{max}} = 0.38 \text{ e \AA}^{-3}$
 4062 reflections $\Delta\rho_{\text{min}} = -0.27 \text{ e \AA}^{-3}$
 272 parameters Extinction correction: *SHELXL97*
 H-atom parameters constrained (Sheldrick, 1997)
 Extinction coefficient: 0.025 (2)

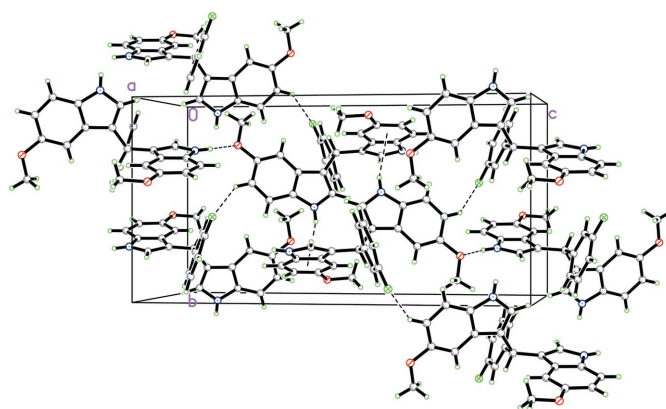


Figure 2
The crystal structure of (I). Dashed lines indicate N–H···O, C–H···F and N–H··· π contacts.

Table 1

Hydrogen-bond geometry (Å, °).

<i>D</i> –H··· <i>A</i>	<i>D</i> –H	H··· <i>A</i>	<i>D</i> ··· <i>A</i>	<i>D</i> –H··· <i>A</i>
N2–H2A···O1 ⁱ	0.86	2.16	2.946 (3)	152
C3–H3A···F ⁱⁱ	0.93	2.51	3.264 (4)	138
N1–H1A···Cg1	0.86	2.46	3.277 (2)	159

Symmetry codes: (i) $x - \frac{1}{2}, -y + \frac{1}{2}, z + \frac{1}{2}$; (ii) $-x + \frac{1}{2}, y + \frac{1}{2}, -z + \frac{1}{2}$.

All H atoms were placed in idealized positions and refined as riding, with C–H = 0.93–0.98 Å and N–H = 0.86 Å, and with $U_{\text{iso}}(\text{H}) = xU_{\text{eq}}(\text{parent atom})$, where $x = 1.5$ for methyl H and 1.2 for all other H atoms.

Data collection: *CAD-4 Software* (Enraf–Nonius, 1989); cell refinement: *CAD-4 Software*; data reduction: *XCAD4* (Harms & Wocadlo, 1995); program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *SHELXTL* (Siemens, 1996); software used to prepare material for publication: *SHELXTL*.

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